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Thermal stability of TiZrAlN films deposited by a reactive magnetron sputtering methodVladimir Uglov¹, Gregory Abadias², Ihar Saladukhin¹, Sergey Zlotski¹¹Belarusian State University, Minsk, Belarus ²Institut P', Poitiers, France

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The aim of the present work is to study the thermal stability, under vacuum and air annealing, of quaternary transition metal nitride films, namely TiZrAlN, with emphasis on the role of Al content on the structure and phase formation. $(\text{Ti,Zr})_{1-x}\text{Al}_x\text{N}$ films with thickness of 300 nm have been deposited onto Si (001) wafers by a reactive unbalanced magnetron sputtering method. Ti, Zr and Al targets were co-sputtered under mixed Ar+N₂ plasma discharges. Varying the RF power of the Al target from 20 to 200 W resulted in a concentration of Al (x_{Al}) in the films to increase from 2.6 to 36.4 at.%, while the Ti:Zr concentration ratio was kept constant to ~ 1.0 . Results of the XRD analysis on as-deposited films indicate the formation of (Ti,Zr,Al)N solid solution. However, the structure changes with rising Al concentration from crystalline to nanocrystalline and then it turns into the amorphous state. Annealing in vacuum ($\sim 10^{-4}$ Pa) at the temperature of 600°C doesn't cause any essential change of (Ti,Zr,Al)N solid solution structure. After annealing in vacuum at 950°C, the structure of the films depends essentially on the aluminum content. At the smaller Al concentrations, the (Ti,Zr,Al)N solid solution remains stable. When $x_{\text{Al}} > 9$ at.% the volume content of ZrN mononitride phase increases and when $x_{\text{Al}} > 13$ at.% the TiN and ZrN phases become dominating. At the same time, Al is apparently included substitutionally into these phases by forming the solid solutions of TiAlN and ZrAlN type. Annealing under air atmosphere of the investigated TiZrAlN samples was carried out for temperature intervals ranging from 400 to 950°C using in situ temperature XRD. Decomposition of the (Ti,Zr,Al)N solid solution already starts at the temperature of 500°C and the diffraction lines corresponding to a solid solution disappear at $T = 600^\circ\text{C}$. The intensive formation of ZrO₂, TiO₂ and Al₂O₃ oxide phases occurs at a temperature of 780°C. Increase in x_{Al} up to 25 at.% leads to oxide formation reduction that can be explained by the passivating role of the Al₂O₃ surface layer. Thus, higher oxidation resistance of TiZrAlN films is reached in a case of high Al concentration at which amorphization of their structure takes place.

Keywordsnanocomposite TiZrAlN
thermal stability
solid solution